

## Synthesis of 1,3-Selenazoles and 2-Imidazolin-5-selones from Isoselenocyanates and Isocyanides

Hajime Maeda,<sup>†</sup> Nobuaki Kambe, and Noboru Sonoda<sup>\*‡</sup>

Department of Applied Chemistry, Faculty of Engineering, Osaka University, Suita, Osaka 565, Japan

Shin-ichi Fujiwara<sup>\*</sup> and Tsutomu Shin-ike

Department of Chemistry, Osaka Dental University, Hirakata, Osaka 573, Japan

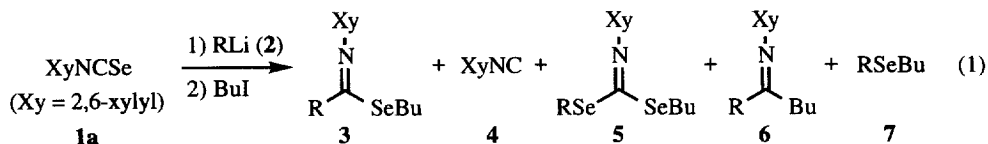
**Abstract:** 1,3-Selenazoles and 2-imidazolin-5-selones were synthesized by the reaction of isoselenocyanates (**1**) with  $\alpha$ -lithiated isocyanides (**11**). Isocyanides having only one substituent at the  $\alpha$ -carbon such as ethyl isocyanoacetate and benzyl isocyanide gave 1,3-selenazoles (**9**) in good yields. On the other hand,  $\alpha,\alpha$ -disubstituted isocyanides such as  $\alpha$ -methylbenzyl isocyanide and diphenylmethyl isocyanide afforded 2-butylseleno-2-imidazolin-5-selones (**18**) after trapping with butyl iodide. The latter products were formed from one molecule of isocyanides and two molecules of isoselenocyanates. Plausible reaction pathways are proposed. © 1997 Elsevier Science Ltd.

### 1. INTRODUCTION

1,3-Selenazoles have attracted much attention not only as useful compounds in dye's chemistry<sup>1</sup> but also in medicinal fields.<sup>2</sup> However, the hitherto known synthetic methods of 1,3-selenazoles are limited only to the reaction of selenoamides or selenoureas with  $\alpha$ -haloketones which provide 1,3-selenazoles having a substituent at the 2-position,<sup>3</sup> and the synthesis of 2-unsubstituted 1,3-selenazoles has been a long-standing theme.<sup>3a,4</sup>

Isocyanates<sup>5</sup> and isothiocyanates<sup>6</sup> have widely been used as building blocks of various heterocycles. For example, 1,3-thiazoles were synthesized in good yields by the reaction of methyl isocyanoacetate with isothiocyanates in the presence of a base such as *t*-BuOK or NaH.<sup>7</sup> As for isoselenocyanates (**1**), their reactions with heteroatom nucleophiles have also been employed for the synthesis of selenium-containing heterocycles,<sup>8-11</sup> however those with carbon nucleophiles are much less studied.<sup>12</sup>

We have recently revealed that the reaction of 2,6-xylyl isoselenocyanate (**1a**) with organolithium compounds (**2**) afforded carbophilic product (**3**) and/or selenophilic products (**4-7**) depending on the nature of **2** (eq 1).<sup>13</sup> For example, phenyllithium attacked the selenium atom exclusively leading to the elimination of isocyanide (**4**), whereas thermodynamically stable organolithiums reacted at the central carbon of **1a** to afford

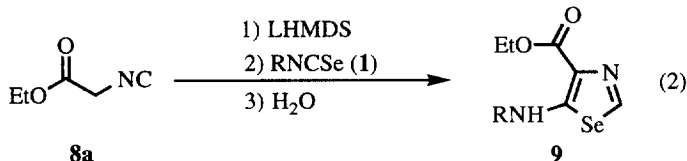


the corresponding lithium selenocarbimidates which were then alkylated by butyl iodide to give selenoimidates (**3**) in good to high yields. These results along with the background as mentioned above prompted us to examine the reaction of isoselenocyanates (**1**) with  $\alpha$ -lithiated isocyanides (**11**)<sup>14</sup> aiming at the preparation of 1,3-selenazoles having no substituent at the 2-position.

## 2. RESULTS AND DISCUSSION

### 2-1. Reaction of Isoselenocyanates with Lithiated $\alpha$ -Monosubstituted Isocyanides

A lithium enolate of ethyl isocyanoacetate was prepared by treating **8a** with 1.1 equiv of lithium hexamethyldisilazide (LHMDS) in THF at  $-78^\circ\text{C}$  for 30 min. To the solution was added 2,6-xylyl isoselenocyanate (**1a**) at  $-78^\circ\text{C}$  and the mixture was stirred at the same temperature for 10 min, and then at  $20^\circ\text{C}$  for 1 h. Quenching with water and usual workup gave 1,3-selenazole (**9a**) in only 11% yield (eq 2, run 1 in Table 1). Addition of 3 equiv of hexamethylphosphoric triamide (HMPA) as an additive improved the yield of **9a** to 46% (run 2). When the reaction time was prolonged up to 5 h the yield was increased, but further prolonged reaction time did not affect the yield (runs 3, 4). The use of *N,N*-dimethylpropyleneurea (DMPU) and *N,N,N,N*-tetramethyl-1,2-ethylenediamine (TMEDA) instead of HMPA were not effective (runs 5, 6). Under similar conditions, phenyl isoselenocyanate (**1b**), diphenylmethyl isoselenocyanate (**1c**), and cyclohexyl isoselenocyanate (**1d**) gave the corresponding 1,3-selenazoles (**9b-d**) in good yields (runs 7-9).



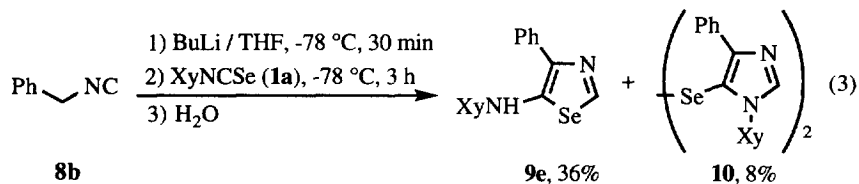
**Table 1.** Reaction of Isoselenocyanates (**1**) with Lithium Enolate of Ethyl Isocyanoacetate (**8a**)

run	RNCSe	additive	time (h)	product	yield (%) <sup>a</sup>
1	XyNCSe ( <b>1a</b> )	none	1	<b>9a</b> (R = Xy)	11
2	<b>1a</b>	HMPA	1	<b>9a</b>	46
3	<b>1a</b>	HMPA	5	<b>9a</b>	58
4	<b>1a</b>	HMPA	15	<b>9a</b>	57
5	<b>1a</b>	DMPU	5	<b>9a</b>	43
6	<b>1a</b>	TMEDA	5	<b>9a</b>	23
7	PhNCSe ( <b>1b</b> )	HMPA	5	<b>9b</b> (R = Ph)	59
8	Ph <sub>2</sub> CHNCSe ( <b>1c</b> )	HMPA	5	<b>9c</b> (R = Ph <sub>2</sub> CH)	73
9	<i>c</i> -C <sub>6</sub> H <sub>11</sub> NCSe ( <b>1d</b> )	HMPA	5	<b>9d</b> (R = <i>c</i> -C <sub>6</sub> H <sub>11</sub> )	71

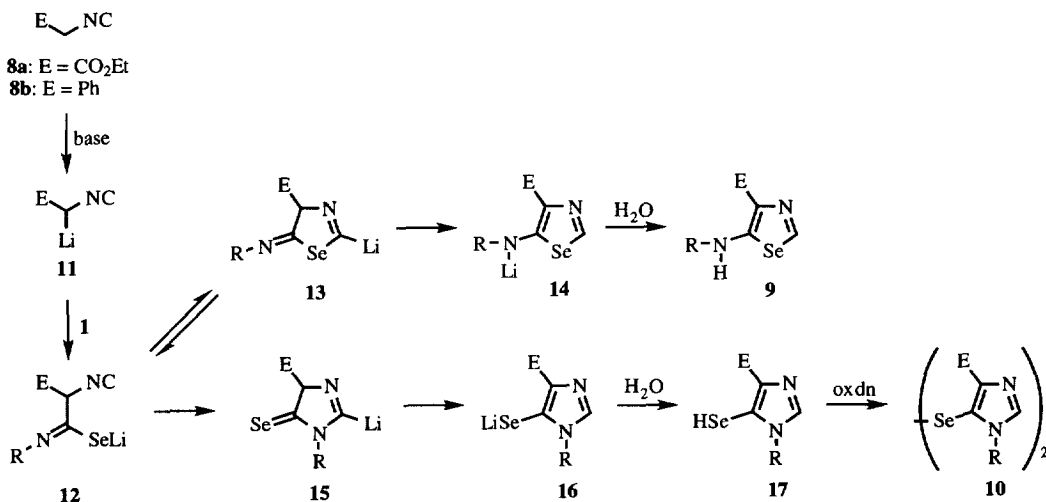
Conditions: **8a** (2.0 mmol), LHMDS (2.2 mmol), THF (25 mL), additive (6.0 mmol),  $-78^\circ\text{C}$ , 30 min; **1** (2.2 mmol),  $-78^\circ\text{C}$ , 10 min, then stirred at  $20^\circ\text{C}$  for the time specified.

a) Isolated yield based on **8a**.

When the reaction of benzyl isocyanide (**8b**) with **1a** was carried out under similar conditions as specified in run 3 of Table 1, a complex mixture was obtained without formation of 1,3-selenazole (**9e**). Then we examined reaction conditions and found that **9e** was formed in 36% yield together with 8% of diimidazolyl diselenide (**10**) when  $\alpha$ -lithiobenzyl isocyanide, prepared from **8b** and BuLi in THF at  $-78\text{ }^{\circ}\text{C}$  for 30 min, was allowed to react with **1a** at  $-78\text{ }^{\circ}\text{C}$  for 3 h without additives (eq 3).



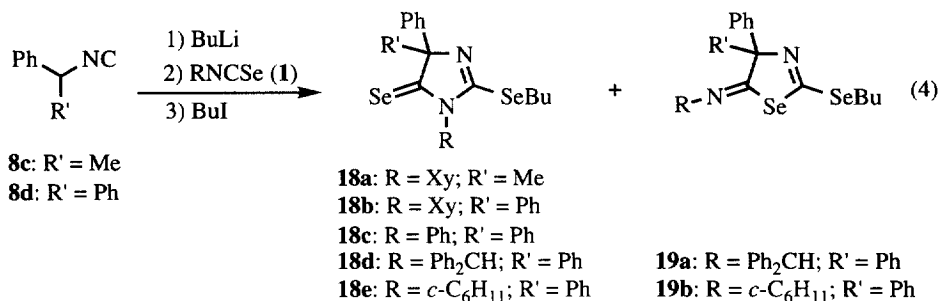
Possible reaction pathways for the formation of 1,3-selenazole (**9**) and diimidazolyl diselenide (**10**) are illustrated in Scheme 1. Reaction of isocyanides (**8**) with organolithium reagents affords  $\alpha$ -lithiated isocyanides (**11**), which react with isoselenocyanates (**1**) in a carbophilic manner to give ambident anions, lithium selenocarbimidates (**12**). Then the selenium atom attacks intramolecularly at the carbon atom of the isocyanide moiety to give **13**. Aromatization of **13** to **14** followed by protonation affords selenazole (**9**). Diselenide (**10**) may be formed *via* intramolecular attack by the nitrogen atom rather than by the selenium to give **15**, which undergoes aromatization, protonation, and oxidation to give diselenide (**10**). The second pathway is only valid in the case of **8b** where ring opening of **13** leading to **12** can compete with proton transfer leading to **14** since benzylic hydrogen of **13** is not sufficiently acidic.



**Scheme 1.** Possible Pathways for the Formation of 1,3-Selenazoles (**9**) and Diimidazolyl Diselenide (**10**) in the Reaction of Isoselenocyanates (**1**) with Lithiated  $\alpha$ -Monosubstituted Isocyanides (**11**)

## 2-2. Reaction of Isoselenocyanates with Lithiated $\alpha,\alpha$ -Disubstituted Isocyanides

Unlike the cases of  $\alpha$ -monosubstituted isocyanides,  $\alpha,\alpha$ -disubstituted ones gave different types of products since they have only one acidic hydrogen and the aromatization process is not possible. The reaction of **1a** with the lithiated derivative of  $\alpha$ -methylbenzyl isocyanide (**8c**) at  $-78\text{ }^{\circ}\text{C}$  for 1 h followed by quenching with water afforded a mixture of several unidentified products. When the reaction was quenched with butyl iodide, 1-xylyl-2-butylseleno-4-methyl-4-phenyl-2-imidazolin-5-selone (**18a**) was obtained in 40% yield based on **8c** (eq 4, run 1 in Table 2). It is noticeable that **18a** includes two selenium atoms. Thus, we carried out a similar reaction using 3 equiv of **1a** and found that the yield of **18a** was drastically increased up to 97% (run 2). In a similar manner, 2-imidazolin-5-selones were also prepared from diphenylmethyl isocyanide (**8d**) by the reaction with **1a**, PhNCSe (**1b**) and Ph<sub>2</sub>CHNCSe (**1c**) (runs 3-5). It should be noted that under the same conditions *c*-C<sub>6</sub>H<sub>11</sub>NCSe (**1d**) afforded **19b** preferentially rather than **18e**, however, the latter was obtained as the major product when the reaction time was prolonged (runs 6, 7).



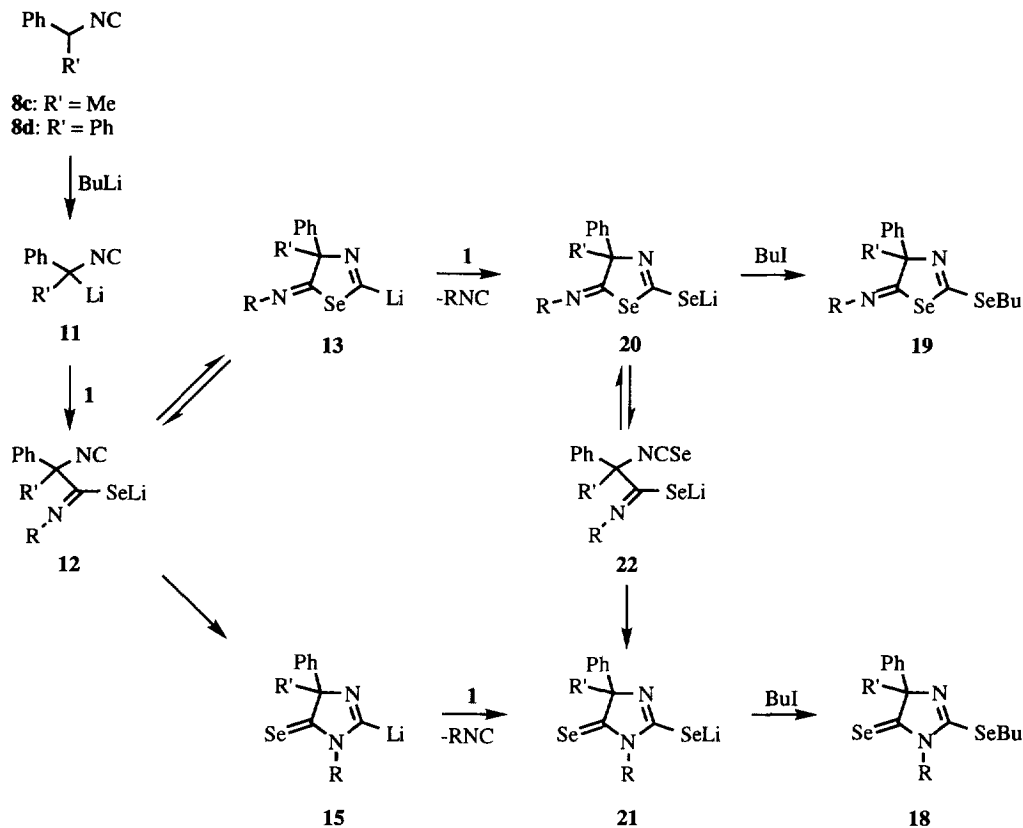
**Table 2.** Reaction of Isoselenocyanates (**1**) with  $\alpha$ -Lithio- $\alpha$ -methylbenzyl Isocyanide (**8c**) or  $\alpha$ -Lithiodiphenylmethyl Isocyanide (**8d**)

run	isocyanide	RNCSe	RNCSe (equiv)	time (h)	yields of products (%) <sup>a</sup>	
					<b>18</b>	<b>19</b>
1	<b>8c</b>	XyNCSe ( <b>1a</b> )	1.2	1	40	0
2	<b>8c</b>	<b>1a</b>	3.0	1	97	0
3	<b>8d</b>	<b>1a</b>	3.0	1	87	0
4	<b>8d</b>	PhNCSe ( <b>1b</b> )	3.0	1	87	0
5	<b>8d</b>	Ph <sub>2</sub> CHNCSe ( <b>1c</b> )	3.0	1	79	7
6	<b>8d</b>	<i>c</i> -C <sub>6</sub> H <sub>11</sub> NCSe ( <b>1d</b> )	3.0	1	14	37
7	<b>8d</b>	<b>1d</b>	3.0	3	72	<1

Conditions: **8** (2.0 mmol), BuLi (2.2 mmol), THF (25 mL),  $-78\text{ }^{\circ}\text{C}$ , 30 min; **1**,  $-78\text{ }^{\circ}\text{C}$ , 30 min; BuI (4.0 mmol),  $-78\text{ }^{\circ}\text{C}$ , 10 min,  $20\text{ }^{\circ}\text{C}$ , 1 h (or 3 h). a) Isolated yields based on **8**.

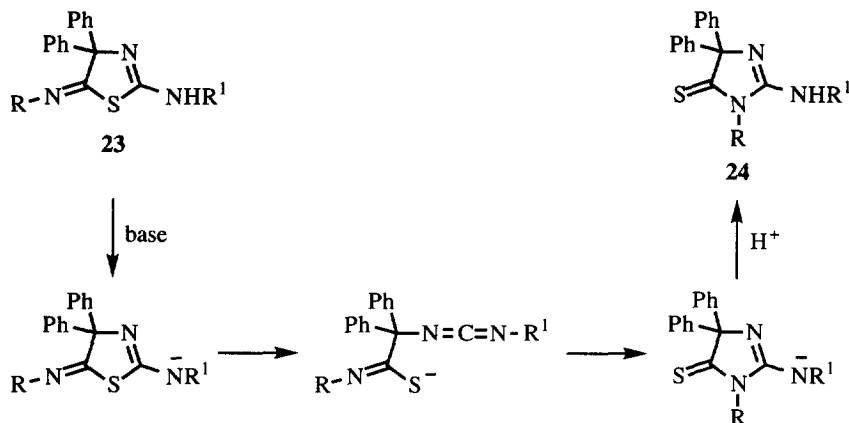
The formation of 2-imidazolin-5-selones (**18**) and 5-imino-2-selenazolines (**19**) could be rationalized as depicted in Scheme 2. The pathway to the intermediate (**13**) is the same as mentioned above, but **13** reacts

with another molecule of isoselenocyanate in a selenophilic manner to give **20** since **13** has no additional acidic hydrogen and aromatization of **13** can not proceed. Thus formed **20** rearranges to **21** via **22** and subsequent alkylation results in the formation of **18**.<sup>15</sup>



**Scheme 2.** Possible Pathways for the Formation of 2-Imidazolin-5-selones (**18**) and 5-Imino-2-selenazolines (**19**) in the Reaction of Isoselenocyanates (**1**) with Lithiated  $\alpha,\alpha$ -Disubstituted Isocyanides (**11**)

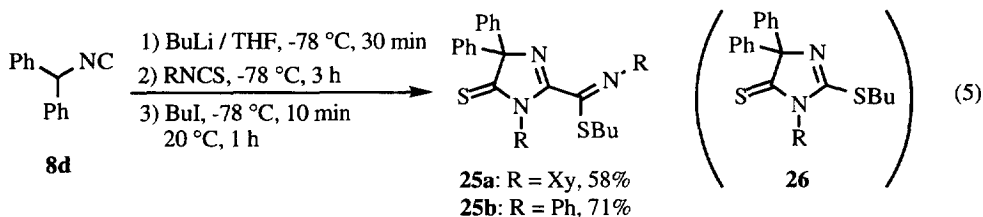
A similar base-induced rearrangement of 2-amino-5-imino-2-thiazolines (**23**) to 2-amino-2-imidazolin-5-thiones (**24**) was examined and it is known that aromatic substituents on the imino nitrogen (R) accelerate the rearrangement in comparison to the cases of aliphatic ones (Scheme 3).<sup>16</sup> This is coincide with our experimental results that **19** were obtained only from aliphatic isoselenocyanates, and in the reaction of **1d** the prolonged reaction time increased the yield of **18e** with suppression of the formation of **19b**. But alternative pathway leading to **21** via **15** can not be ruled out.



**Scheme 3.** Base-Induced Rearrangement of **23** to **24**

The result that  $\alpha$ -lithiated isocyanides (**11**) attack the central carbon of **1** whereas 2-lithio-2-selenazoles (**13**) and 2-lithio-2-imidazolines (**15**) attack the selenium atom of **1** does not contradict our recent finding that phenyllithium, which is an  $sp^2$  anion like **13** and **15**, react with isoselenocyanates in a selenophilic manner and thermodynamically stable carbanions afforded carbophilic products predominantly.<sup>13</sup>

In order to compare the site selectivities of **1** with its sulfur analogues, we carried out the reaction of 2,6-xylyl and phenyl isothiocyanates under the same conditions. Interestingly, only 2-imidazolin-5-thione-2-thiocarboximidates (**25**) were obtained without thiophilic products **26** (eq 5).<sup>17</sup>



### 3. CONCLUSION

We have established a convenient synthetic method of 1,3-selenazoles having no substituent at the 2-position by the reaction of isoselenocyanates with lithiated  $\alpha$ -monosubstituted isocyanides. The similar reaction of isoselenocyanates with lithiated  $\alpha,\alpha$ -disubstituted isocyanides followed by trapping with butyl iodide afforded 2-butylseleno-2-imidazolin-5-selones and 2-butylseleno-5-imino-2-selenazoles. There has been reported only one preparative method for each class of compounds.<sup>18,19</sup> The reactions described herein would provide the efficient routes to these selenium-containing heterocycles.

## EXPERIMENTAL SECTION

### General Comments

THF was distilled from sodium benzophenone ketyl. HMPA, DMPU, TMEDA, and hexamethyldisilazane were fractionally distilled and dried over calcium hydride. BuI was distilled from  $P_2O_5$ . BuLi, ethyl isocyanoacetate (**8a**), and benzyl isocyanide (**8b**) were used as purchased.  $\alpha$ -Methylbenzyl isocyanide (**8c**) and diphenylmethyl isocyanide (**8d**) were prepared according to the literature,<sup>20</sup> and purified by distillation and recrystallization, respectively. Isoselenocyanates (**1a-d**)<sup>21</sup> and 2,6-xylyl isothiocyanate<sup>22</sup> were synthesized by the reported procedures, and purified by silica gel column chromatography. Phenyl isothiocyanate was commercially available, and was purified by distillation.

Melting points were determined on a Yanagimoto Micro Melting Point apparatus.  $^1H$  and  $^{13}C$  NMR spectra were recorded on a JEOL JNM-GSX-270 (270 MHz and 68 MHz, respectively) or a JEOL JNM-ALICE-400 (400 MHz and 100 MHz, respectively) spectrometer using  $Me_4Si$  as an internal standard. IR spectra were determined on a Perkin Elmer Model 1600 spectrometer. Purification of products was performed on a recycling preparative HPLC (Japan Analytical Industry Co. Ltd., Model LC-908) equipped with JAIGEL-1H and -2H columns (GPC) using  $CHCl_3$  as an eluent or by column chromatography using Fuji-Davison silica gel WB-300 (100-250 mesh) or by preparative TLC with Wakogel B-5F silica gel (325 mesh). Sufficiently pure products were obtained by these procedures and recrystallization was not needed. Mass spectra (EI) were taken on a SHIMADZU GCMC-QP2000 operating in the electron impact mode (70 eV) equipped with CBP1-M25-025 column. Mass spectra (CI or FAB) were obtained on a JEOL JMS-DX303 in the Instrumental Analysis Center of the Faculty of Engineering, Osaka University. Elemental analyses were performed on Perkin Elmer 240C apparatus.

### Reaction of Lithium Enolate of **8a** with **1a**

Ethyl isocyanoacetate (**8a**, 217 mg, 1.92 mmol) was added at  $-78^\circ C$  to the solution of LHMDs, generated by the reaction of hexamethyldisilazane (496 mg, 3.07 mmol) and BuLi (1.64 M in hexane, 1.4 mL, 2.30 mmol) in THF (25 mL) / HMPA (1 mL) at  $-78^\circ C$ , and the mixture was stirred for 30 min. To the mixture was added  $XyNCSe$  (**1a**, 484 mg, 2.30 mmol) at the same temperature, and the stirring was continued for 10 min. The mixture was then warmed up to  $20^\circ C$  and stirred for 5 h. Aqueous saturated NaCl solution (50 mL) was added, and the product was extracted with ether (50 mL), dried over  $MgSO_4$ , and concentrated. The residue was purified by recycling preparative HPLC to afford 4-ethoxycarbonyl-5-(2,6-dimethylphenyl)amino-1,3-selenazole (**9a**, 357 mg, 58% yield based on **8a**). White solid; mp  $126^\circ C$ ;  $^1H$  NMR (270 MHz,  $CDCl_3$ )  $\delta$  1.45 (t,  $J = 7.1$  Hz, 3 H), 2.29 (s, 6 H), 4.43 (q,  $J = 7.1$  Hz, 2 H), 7.12-7.21 (m, 3 H), 8.71 (d,  $J = 1.0$  Hz, 1 H), 9.06 (brs, 1 H);  $^{13}C$  NMR (68 MHz,  $CDCl_3$ )  $\delta$  14.64, 17.88, 60.55, 121.26, 128.29, 129.12, 135.88, 138.44, 141.14, 165.47, 169.93; IR (KBr) 2362, 1660, 1530, 1410, 1380, 1232, 1172  $cm^{-1}$ ; MS (CI),  $m/z$  (%) = 105 (2), 279 (23), 325 ( $M^+ + 1$ , 100). Anal. Calcd for  $C_{14}H_{16}N_2O_2Se$ : C, 52.02; H, 4.99; N, 8.67. Found: C, 52.10; H, 5.03; N, 8.57.

**4-Ethoxycarbonyl-5-phenylamino-1,3-selenazole (9b)**

Purified by recycling preparative HPLC (59% yield based on **8a**). White solid; mp 64–65 °C;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  1.46 (t,  $J = 7.1$  Hz, 3 H), 4.45 (q,  $J = 7.1$  Hz, 2 H), 7.15 (t,  $J = 7.2$  Hz, 1 H), 7.27 (d,  $J = 7.2$  Hz, 2 H), 7.39 (t,  $J = 7.2$  Hz, 2 H), 8.85 (s, 1 H), 10.27 (brs, 1 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  14.52, 60.74, 118.10, 124.03, 124.14, 129.41, 138.34, 141.40, 161.43, 165.16; IR (KBr) 1660, 1538, 1413, 1377, 1243, 1199, 1173, 758  $\text{cm}^{-1}$ ; MS (EI),  $m/e$  (%) = 77 (22), 104 (12), 169 (11), 223 (4), 250 (100), 296 ( $\text{M}^+$ , 60). Anal. Calcd for  $\text{C}_{12}\text{H}_{12}\text{N}_2\text{O}_2\text{Se}$ : C, 48.83; H, 4.10; N, 9.49. Found: C, 48.80; H, 3.94; N, 9.69.

**4-Ethoxycarbonyl-5-(diphenylmethyl)amino-1,3-selenazole (9c)**

Purified by recycling preparative HPLC (73% yield based on **8a**). White solid; mp 96–97 °C;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  1.40 (t,  $J = 7.1$  Hz, 3 H), 4.38 (q,  $J = 7.1$  Hz, 2 H), 5.35 (d,  $J = 5.1$  Hz, 1 H), 7.25–7.40 (m, 10 H), 8.71 (brs, 2 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  14.55, 60.33, 69.26, 121.32, 127.05, 127.82, 128.65, 138.71, 139.45, 164.86, 167.00; IR (KBr) 3301, 3041, 2973, 1652, 1530, 1410, 1238, 1179  $\text{cm}^{-1}$ ; MS (EI),  $m/e$  (%) = 167 (100), 386 ( $\text{M}^+$ , 18). Anal. Calcd for  $\text{C}_{19}\text{H}_{18}\text{N}_2\text{O}_2\text{Se}$ : C, 59.23; H, 4.71; N, 7.27. Found: C, 59.20; H, 4.66; N, 7.13.

**4-Ethoxycarbonyl-5-cyclohexylamino-1,3-selenazole (9d)**

Purified by recycling preparative HPLC (71% yield based on **8a**). Yellow oil;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  1.22–1.50 (m, 5 H), 1.41 (t,  $J = 7.1$  Hz, 3 H), 1.61 (brs, 1 H), 1.77 (brs, 2 H), 2.06 (brs, 2 H), 2.94 (brs, 1 H), 4.37 (q,  $J = 7.1$  Hz, 2 H), 8.03 (brs, 1 H), 8.71 (s, 1 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  14.61, 24.34, 25.27, 32.36, 60.08, 61.55, 119.91, 136.66, 164.97, 167.24; IR (NaCl) 3292, 2932, 2856, 1655, 1537, 1411, 1225, 1177, 1144, 732  $\text{cm}^{-1}$ ; MS (EI),  $m/e$  (%) = 83 (20), 302 ( $\text{M}^+$ , 100). HRMS Calcd for  $\text{C}_{12}\text{H}_{18}\text{N}_2\text{O}_2\text{Se}$ : 302.0534. Found: 302.0534.

**Reaction of Lithio Derivative of **8b** with **1a****

BuLi (1.69 M in hexane, 1.4 mL, 2.37 mmol) was added to the solution of benzyl isocyanide (**8b**, 231 mg, 1.97 mmol) in THF (25 mL) at  $-78$  °C, and the mixture was stirred for 30 min. To the solution was added XyNCSe (**1a**, 472 mg, 2.25 mmol) at the same temperature, and the mixture was stirred for 3 h. Aqueous saturated NaCl solution (50 mL) was added at  $-78$  °C, and the product was extracted with ether (50 mL), dried over  $\text{MgSO}_4$ , and concentrated. The residue was purified by recycling preparative HPLC to afford 4-phenyl-5-(2,6-dimethylphenyl)amino-1,3-selenazole (**9e**, 230 mg, 36% yield based on **8b**) and 1,1'-di(2,6-diphenylmethyl)-4,4'-diphenyl-2,2'-diimidazolyl diselenide (**10**, 52 mg, 8% yield based on **8b**). **Data for 9e**. White solid; mp 149 °C;  $^1\text{H}$  NMR (270 MHz,  $\text{CDCl}_3$ )  $\delta$  2.30 (s, 6 H), 5.68 (brs, 1 H), 7.08–7.13 (m, 3 H), 7.29 (t,  $J = 7.3$  Hz, 1 H), 7.46 (t,  $J = 7.8$  Hz, 2 H), 7.97 (d,  $J = 7.3$  Hz, 2 H), 9.12 (s, 1 H);  $^{13}\text{C}$  NMR (68 MHz,  $\text{CDCl}_3$ )  $\delta$  18.18, 126.43, 126.67, 127.40, 128.86, 129.18, 134.05, 135.17, 135.59, 143.22, 144.30, 152.20; IR (KBr) 3256, 2360, 1526, 1448, 1439, 776, 734, 700  $\text{cm}^{-1}$ ; MS (EI),  $m/e$  (%) = 77 (11), 105 (11), 132 (17), 220 (33), 328 ( $\text{M}^+$ , 100). Anal. Calcd for  $\text{C}_{17}\text{H}_{16}\text{N}_2\text{Se}$ : C, 62.39; H, 4.93; N, 8.56. Found: C, 62.22; H, 4.85; N, 8.49. **Data for 10**. Yellow solid; mp 199–200 °C;  $^1\text{H}$  NMR (270 MHz,  $\text{CDCl}_3$ )  $\delta$  1.93 (s, 12 H), 7.10 (d,  $J = 7.5$  Hz, 4 H), 7.10–7.28 (m, 8 H), 7.65 (s, 2 H), 7.85 (d,  $J = 7.8$  Hz, 4 H);  $^{13}\text{C}$  NMR (68 MHz,  $\text{CDCl}_3$ )  $\delta$  18.09, 111.47, 127.77, 127.80, 127.97, 128.32, 129.41, 133.39, 134.68, 136.48, 140.28, 148.89; IR (KBr) 1485, 773,

693, 668  $\text{cm}^{-1}$ ; MS (EI),  $m/e$  (%) = 77 (8), 105 (6), 220 (22), 247 (19), 327 (100), 654 ( $M^+$ , 5). HRMS Calcd for  $C_{34}H_{30}N_4Se_2$ : 654.0801. Found: 654.0790.

#### Reaction of Lithio Derivative of **8c** with **1a**

BuLi (1.70 M in hexane, 1.3 mL, 2.21 mmol) was added to the solution of  $\alpha$ -methylbenzyl isocyanide (**8c**, 263 mg, 2.00 mmol) in THF (25 mL) at  $-78^\circ\text{C}$ , and the mixture was stirred for 30 min. To the solution was added XyNCSe (**1a**, 1258 mg, 5.99 mmol) at the same temperature, and the mixture was stirred for 1 h. After BuI (782 mg, 4.25 mmol) was added at  $-78^\circ\text{C}$ , the stirring was continued for 10 min, and then at  $20^\circ\text{C}$  for 1 h. Aqueous saturated  $\text{NH}_4\text{Cl}$  solution (50 mL) was added, and the product was extracted with ether (50 mL), dried over  $\text{MgSO}_4$ , and concentrated. The residue was purified by silica gel column chromatography (hexane/ether = 5/1) to afford 1-(2,6-dimethylphenyl)-2-butylseleno-4-methyl-4-phenyl-2-imidazolin-5-selone (**18a**, 924 mg, 97% based on **8c**). Yellow oil;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  0.93 (t,  $J = 7.3$  Hz, 3 H), 1.43 (sext,  $J = 7.3$  Hz, 2 H), 1.80 (quint,  $J = 7.3$  Hz, 2 H), 1.98 (s, 3 H), 2.07 (s, 3 H), 2.19 (s, 3 H), 3.28 (t,  $J = 7.3$  Hz, 2 H), 7.16 (dd,  $J = 7.4, 4.4$  Hz, 2 H), 7.27–7.39 (m, 4 H), 7.53 (d,  $J = 7.3$  Hz, 2 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  13.56, 17.57, 17.73, 22.81, 27.52, 27.76, 31.63, 91.83, 126.23, 127.76, 128.37, 128.77, 130.26, 133.54, 136.36, 138.99, 158.42, 220.95; IR (NaCl) 2958, 2929, 1586, 1574, 1568, 1352, 1260, 1239, 1172, 1134, 768, 696  $\text{cm}^{-1}$ ; MS (FAB),  $m/z$  (%) = 57 (8), 77 (19), 91 (14), 105 (34), 130 (27), 210 (53), 235 (100), 342 (19), 398 (25), 479 ( $M^+ + 1$ , 88). HRMS Calcd for  $C_{22}H_{26}N_2Se_2$ : 478.0427. Found: 478.0435.

#### 1-(2,6-Dimethylphenyl)-2-butylseleno-4,4-diphenyl-2-imidazolin-5-selone (**18b**)

Purified by silica gel column chromatography (hexane/ether = 3/1) (87% yield based on **8d**). Orange oil;  $^1\text{H}$  NMR (270 MHz,  $\text{CDCl}_3$ )  $\delta$  0.90 (t,  $J = 7.3$  Hz, 3 H), 1.40 (sext,  $J = 7.3$  Hz, 2 H), 1.78 (quint,  $J = 7.3$  Hz, 2 H), 2.11 (s, 6 H), 3.28 (t,  $J = 7.3$  Hz, 2 H), 7.17 (d,  $J = 7.3$  Hz, 2 H), 7.27–7.38 (m, 7 H), 7.55–7.62 (m, 4 H);  $^{13}\text{C}$  NMR (68 MHz,  $\text{CDCl}_3$ )  $\delta$  13.46, 17.71, 22.68, 27.50, 31.65, 97.38, 127.48, 127.89, 128.27, 128.64, 130.14, 133.59, 136.19, 140.33, 158.06, 216.64; IR (NaCl) 2957, 2929, 1586, 1574, 1568, 1446, 1347, 1259, 1237, 1147, 771, 760, 697  $\text{cm}^{-1}$ ; MS (CI),  $m/z$  (%) = 167 (4), 298 (5), 405 (6), 461 (10), 485 (11), 541 ( $M^+ + 1$ , 100). HRMS Calcd for  $C_{27}H_{28}N_2Se_2$ : 540.0583. Found: 540.0571.

#### 1,4,4-Triphenyl-2-butylseleno-2-imidazolin-5-selone (**18c**)

Purified by recycling preparative HPLC (87% yield based on **8d**). Orange solid; mp  $122^\circ\text{C}$ ;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  0.89 (t,  $J = 7.4$  Hz, 3 H), 1.39 (sext,  $J = 7.4$  Hz, 2 H), 1.78 (quint,  $J = 7.4$  Hz, 2 H), 3.27 (t,  $J = 7.4$  Hz, 2 H), 7.28–7.36 (m, 8 H), 7.50–7.54 (m, 3 H), 7.57–7.62 (m, 4 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  13.49, 22.84, 28.38, 31.59, 96.92, 127.30, 127.66, 127.83, 128.04, 129.35, 129.76, 136.24, 139.85, 157.41, 218.51; IR (KBr) 1585, 1568, 1359, 1240, 1145, 691  $\text{cm}^{-1}$ ; MS (EI),  $m/e$  (%) = 77 (12), 165 (90), 192 (14), 272 (59), 376 (100), 432 (15), 456 (9), 512 ( $M^+$ , 25). Anal. Calcd for  $C_{25}H_{24}N_2Se_2$ : C, 58.83; H, 4.74; N, 5.49. Found: C, 58.81; H, 4.73; N, 5.59.

#### Reaction of Lithio Derivative of **8d** with **1c**

Products were obtained from diphenylmethyl isocyanide (**8d**) and diphenylmethyl isoselenocyanate (**1c**) in a similar manner as in the case of **8c**. The residue was purified by recycling preparative HPLC followed by

PTLC (hexane/ether = 30/1) to afford 1-diphenylmethyl-2-butylseleno-4,4-diphenyl-2-imidazolin-5-selone (**18d**, 79% based on **8d**) and 2-butylseleno-4,4-diphenyl-5-(*N*-diphenylmethyl)imino-2-selenazoline (**19a**, 7% yield based on **8d**). **Data for 18d**. Yellow solid; mp 138 °C; <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>) δ 0.80 (t, *J* = 7.3 Hz, 3 H), 1.26 (sext, *J* = 7.3 Hz, 2 H), 1.58 (quint, *J* = 7.3 Hz, 2 H), 3.08 (t, *J* = 7.3 Hz, 2 H), 7.23–7.48 (m, 20 H), 7.79 (s, 1 H); <sup>13</sup>C NMR (68 MHz, CDCl<sub>3</sub>) δ 13.52, 22.89, 29.54, 31.43, 65.58, 96.37, 127.57, 127.97, 128.40, 128.49, 129.10, 135.96, 140.65, 156.71, 219.53; IR (KBr) 2954, 2859, 1566, 1492, 1446, 1312, 1220, 1130, 696 cm<sup>-1</sup>; MS (FAB), *m/z* (%) = 167 (100), 208 (23), 273 (37), 522 (7), 603 (*M*<sup>+</sup>+1, 43). Anal. Calcd for C<sub>32</sub>H<sub>30</sub>N<sub>2</sub>Se<sub>2</sub>: C, 64.00; H, 5.04; N, 4.66. Found: C, 63.67; H, 5.04; N, 4.54. **Data for 19a**. White solid; mp 145 °C; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 0.89 (t, *J* = 7.4 Hz, 3 H), 1.40 (sext, *J* = 7.4 Hz, 2 H), 1.78 (quint, *J* = 7.4 Hz, 2 H), 3.25 (t, *J* = 7.4 Hz, 2 H), 4.75 (s, 1 H), 7.12–7.30 (m, 16 H), 7.33–7.40 (m, 4 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 13.51, 22.91, 27.61, 32.43, 83.76, 96.86, 126.86, 126.96, 127.06, 127.40, 127.74, 127.94, 141.75, 142.35, 149.31, 173.93; IR (KBr) 2929, 1668, 1590, 1492, 1447, 784, 697 cm<sup>-1</sup>; MS (CI), *m/z* (%) = 167 (100), 195 (13), 272 (15), 329 (15), 603 (*M*<sup>+</sup>+1, 36). Anal. Calcd for C<sub>32</sub>H<sub>30</sub>N<sub>2</sub>Se<sub>2</sub>: C, 64.00; H, 5.04; N, 4.66. Found: C, 63.86; H, 5.19; N, 4.56.

#### Reaction of Lithio Derivative of **8d** with **1d**

Products were obtained from diphenylmethyl isocyanide (**8d**) and cyclohexyl isoselenocyanate (**1d**) in a similar manner as in the case of **8c**. The residue was purified by recycling preparative HPLC then by PTLC (hexane/ether = 20/1) to afford 1-cyclohexyl-2-butylseleno-4,4-diphenyl-2-imidazolin-5-selone (**18e**, 14% yield based on **8d**) and 2-butylseleno-4,4-diphenyl-5-(*N*-cyclohexyl)imino-2-selenazoline (**19b**, 37% yield based on **8d**). **Data for 18e**. Yellow solid; mp 126.5–127.5 °C; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 0.89 (t, *J* = 7.3 Hz, 3 H), 1.20–1.52 (m, 4 H), 1.41 (sext, *J* = 7.3 Hz, 2 H), 1.68–1.95 (m, 6 H), 1.77 (quint, *J* = 7.3 Hz, 2 H), 2.19 (brs, 1 H), 3.32 (t, *J* = 7.3 Hz, 2 H), 5.30 (brs, 1 H), 7.23–7.33 (m, 6 H), 7.39–7.50 (m, 4 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 13.52, 22.88, 24.99, 26.00, 29.39, 29.81, 31.55, 59.71, 96.06, 127.10, 127.46, 128.04, 139.94, 154.37, 218.13; IR (KBr) 2942, 2858, 1565, 1446, 1343, 1307, 1208, 1122, 1044, 758, 698 cm<sup>-1</sup>; MS (EI), *m/e* (%) = 165 (51), 193 (58), 246 (43), 272 (100), 382 (62), 438 (12), 462 (6), 518 (*M*<sup>+</sup>, 46). HRMS Calcd for C<sub>25</sub>H<sub>30</sub>N<sub>2</sub>Se<sub>2</sub>: 518.0739. Found: 518.0750. **Data for 19b**. Yellow oil; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 0.90 (t, *J* = 7.3 Hz, 3 H), 1.21–1.38 (m, 3 H), 1.42 (sext, *J* = 7.3 Hz, 2 H), 1.45–1.52 (m, 3 H), 1.52–1.80 (m, 4 H), 1.80 (quint, *J* = 7.3 Hz, 2 H), 2.42 (brs, 1 H), 3.26 (t, *J* = 7.3 Hz, 2 H), 7.18–7.34 (m, 6 H), 7.36–7.50 (m, 4 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 13.54, 22.94, 24.08, 25.51, 27.46, 32.15, 32.48, 77.33, 95.42, 126.85, 127.39, 127.56, 142.56, 149.41, 168.39; IR (NaCl) 2929, 2854, 1667, 1588, 1574, 1446, 906, 883, 803, 778, 759, 740, 696, 656 cm<sup>-1</sup>; MS (CI), *m/z* (%) = 83 (6), 167 (8), 193 (7), 272 (100), 329 (45), 356 (28), 519 (*M*<sup>+</sup>+1, 88). Anal. Calcd for C<sub>25</sub>H<sub>30</sub>N<sub>2</sub>Se<sub>2</sub>: C, 58.14; H, 5.86; N, 5.42. Found: C, 58.30; H, 5.88; N, 5.18.

#### Reaction of Lithio Derivative of **8d** with 2,6-Xylyl Isothiocyanate

The reaction was carried out as in the case of **8c** using diphenylmethyl isocyanide (**8d**) and 2,6-xylyl isothiocyanate, but the reaction time was prolonged to 3 h. The residue was purified by recycling preparative HPLC to afford *S*-butyl *N*-(2,6-dimethylphenyl)-1-(2,6-dimethylphenyl)-4,4-diphenyl-2-imidazolin-5-thione-2-thiocarboximidate (**25a**, 58% yield based on **8d**). Yellow solid; mp 48–50 °C; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 0.63 (t, *J* = 7.4 Hz, 3 H), 0.89 (sext, *J* = 7.4 Hz, 2 H), 1.27 (quint, *J* = 7.4 Hz, 2 H), 1.65 (s, 6 H), 2.14 (s, 6

H), 2.70 (t,  $J = 7.4$  Hz, 2 H), 6.86-6.95 (m, 3 H), 7.10 (d,  $J = 7.6$  Hz, 2 H), 7.21 (t,  $J = 7.6$  Hz, 1 H), 7.30-7.40 (m, 6 H), 7.59 (d,  $J = 8.3$  Hz, 4 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  13.31, 16.86, 18.12, 21.46, 31.66, 32.71, 92.01, 123.87, 125.13, 127.43, 127.54, 127.77, 127.95, 128.31, 129.23, 133.93, 136.24, 140.62, 145.95, 154.70, 154.86, 212.43; IR (KBr) 3060, 2957, 2871, 1601, 1588, 1272, 1026, 894, 764, 698  $\text{cm}^{-1}$ ; MS (EI),  $m/e$  (%) = 57 (4), 77 (3), 105 (5), 131 (10), 192 (6), 329 (33), 355 (3), 486 (54), 518 (100), 560 (52), 575 ( $\text{M}^+$ , 21). Anal. Calcd for  $\text{C}_{36}\text{H}_{37}\text{N}_3\text{S}_2$ : C, 75.09; H, 6.48; N, 7.30. Found: C, 75.00; H, 6.50; N, 7.15.

#### **S-Butyl N-phenyl-1,4,4-triphenyl-2-imidazolin-5-thione-2-thiocarboximidate (25b)**

The reaction was carried out as in the case of **8c** using diphenylmethyl isocyanide (**8d**) and phenyl isothiocyanate, but the reaction time was prolonged to 3 h. Purified by recycling preparative HPLC (71% yield based on **8d**). Yellow solid obtained as a mixture of stereoisomers (major/minor = 72/28); mp 41-42  $^{\circ}\text{C}$ ;  $^1\text{H}$  NMR (270 MHz,  $\text{CDCl}_3$ )  $\delta$  0.62-0.70 (m, 3 H, minor), 0.85-0.95 (m, 2 H, minor), 0.93 (t,  $J = 7.3$  Hz, 3 H, major), 1.20-1.35 (m, 2 H, minor), 1.44 (sext,  $J = 7.3$  Hz, 2 H, major), 1.67 (quint,  $J = 7.3$  Hz, 2 H, major), 2.58-2.67 (m, 2 H, minor), 3.14 (t,  $J = 7.3$  Hz, 2 H, major), 6.32 (d,  $J = 7.8$  Hz, 2 H, major), 6.45-6.52 (m, 2 H, minor), 6.75 (d,  $J = 6.8$  Hz, 2 H, major), 6.88-7.63 (m, major 16 H + minor 18 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  13.34 (minor), 13.58 (major), 21.34 (minor), 21.89 (major), 30.13 (major), 30.20 (major), 31.67 (minor), 32.05 (minor), 92.06 (major), 92.12 (minor), 118.69, 121.05, 124.83, 126.71, 127.59, 127.79, 127.96, 128.23, 128.27, 128.54, 128.65, 129.01, 133.59, 135.34, 140.13, 140.45, 147.22, 147.75, 155.09, 155.61, 156.63, 212.97 (major), 213.11 (minor); IR (KBr) 2957, 2364, 1577, 1492, 1345, 1261, 1034, 755, 694  $\text{cm}^{-1}$ ; MS (EI),  $m/e$  (%) = 57 (17), 77 (13), 103 (2), 136 (75), 192 (68), 210 (34), 224 (20), 295 (6), 327 (10), 430 (81), 462 (45), 519 ( $\text{M}^+$ , 100). HRMS Calcd for  $\text{C}_{32}\text{H}_{29}\text{N}_3\text{S}_2$ : 519.1803. Found: 519.1793.

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<sup>1</sup>Present address: Department of Chemistry, College of Engineering, Osaka Prefecture University, Sakai, Osaka 599, Japan

<sup>3</sup>Present address: Department of Applied Chemistry, Faculty of Engineering, Kansai University, Suita, Osaka 564, Japan

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